

How much of what and how  
often...

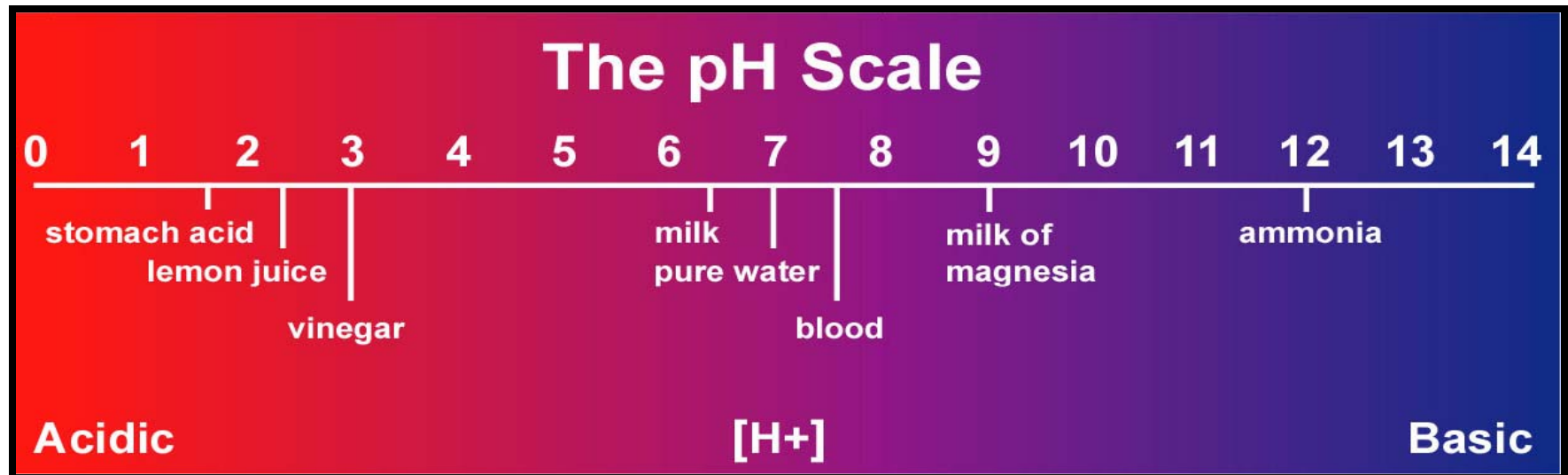
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# What about pH?

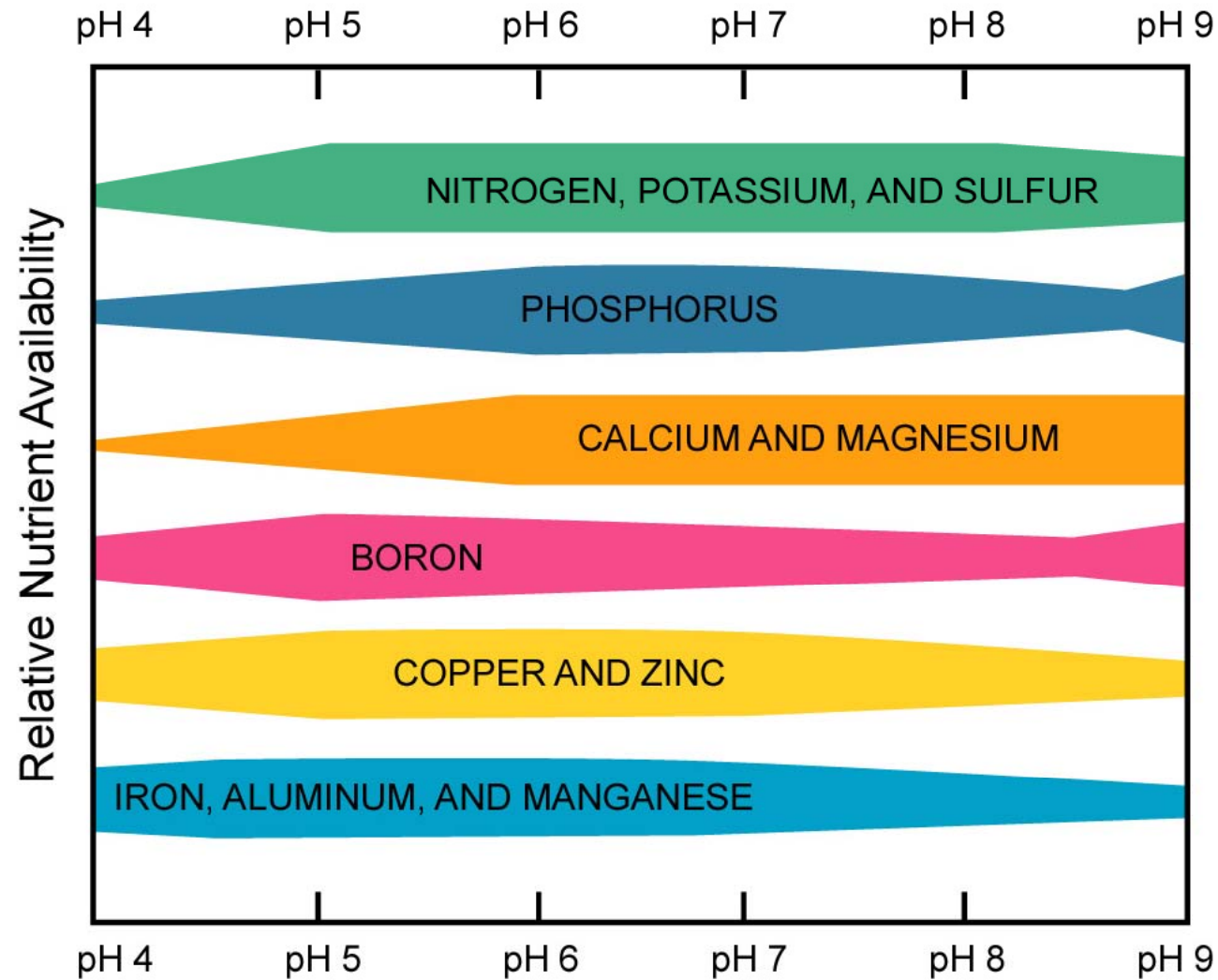
- Generally speaking, for pastures, if pH is in the 6-8 range it is not economical to adjust it
- When considering pH the unit of currency is CaCo<sub>3</sub> Equivalents

# pH

- Indicates relative acidity or alkalinity
- pH 7 = neutral; less than 7 = acid; more than 7 = alkaline or basic



Adapted from library.thinkquest.org



# Soil pH and nutrients

Adapted from [www.soil.ncsu.edu](http://www.soil.ncsu.edu)

Widest part of the bar indicates maximum availability

## Common liming materials include:

Table 2. Common liming materials.

| Name            | Chemical formula                              | (CaCO <sub>3</sub> ) equivalent (%) | Source                   |
|-----------------|---|-------------------------------------|--------------------------|
| Burned lime     | CaO   | 150–175                             | Kiln burned              |
| Hydrated lime   | Ca(OH) <sub>2</sub>                           | 120–135                             | Steam burned             |
| Dolomite*       | CaCO <sub>3</sub> , MgCO <sub>3</sub>         | 110                                 | Natural deposit          |
| Limestone       | CaCO <sub>3</sub>                             | 100                                 | Pure form, finely ground |
| Sugar beet lime | CaCO <sub>3</sub>                             | 80–90                               | Sugar beet byproduct     |
| Power plant ash | CaO, K <sub>2</sub> O, MgO, Na <sub>2</sub> O | 5–50                                | Wood-fired power plants  |

Source: *Western Fertilizer Handbook* (1995), p. 219.

\*High-magnesium dolomite should not be applied where magnesium exceeds calcium in soil.

## How much material needs to be added to change pH?

Table 3. Amount of lime needed to change soil pH (soil reaction).

| Change in pH desired | Tons of lime (100% of CaCO <sub>3</sub> equiv.) required per acre (6-inch depth) for various soil textures* |            |      |           |           |      |
|----------------------|---|------------|------|-----------|-----------|------|
|                      | Sand  | Sandy loam | Loam | Silt loam | Clay loam | Muck |
| 4.0 to 6.5           | 1.3   | 2.5        | 3.5  | 4.2       | 5         | 9.5  |
| 4.5 to 6.5           | 1.1   | 2.1        | 2.9  | 3.5       | 4.2       | 8.1  |
| 5.0 to 6.5           | 0.9   | 1.7        | 2.3  | 2.8       | 3.3       | 6.3  |
| 5.5 to 6.5           | 0.6   | 1.3        | 1.7  | 2         | 2.3       | 4.3  |
| 6.0 to 6.5           | 0.3   | 0.7        | 0.9  | 1.1       | 1.2       | 2.2  |

Source: *Western Fertilizer Handbook* (1995), p. 218.

\* This table gives the amount of 100% CaCO<sub>3</sub> equivalent required to change the soil pH of a 6-inch layer of soil. To alter the pH of the surface 12 inches of soil, twice the amount of lime should be applied.

**Figure 2. Liming rate calculation.**

**Step 1.** These steps determine the amount of ash to apply as a liming material to neutralize soil acidity and raise soil pH. First, obtain the lime ( $\text{CaCO}_3$ ) equivalent from the ash analysis (table 1). We will assume 11% for our example.

**Step 2.** From the soil test report, obtain the CEC and estimate the soil texture. In our example, the CEC is 6, which indicates that the soil has a sand or loamy sand texture (table 4). An alternative way to determine soil texture is to obtain the SP from the soil test report (table 4).

**Step 3.** Use Table 3 to determine the amount of 100%  $\text{CaCO}_3$  equivalent lime needed to raise the soil pH to the desired level by using the soil pH and texture. Obtain the soil pH from the soil test report. In our example, assume the pH is 4.6. To change the pH from 4.6 to 6.5 for a soil of sand texture, an application of 1.1 tons/acre of 100% limestone is required.

**Step 4.** Determine the amount of ash to apply. Divide the amount of 100% limestone by the percentage  $\text{CaCO}_3$  equivalent of the ash. In our example, divide 1.1 from step 3 by 0.11 (the  $\text{CaCO}_3$  equivalent of this ash) from step 1. We find 10.0 tons/acre of dry ash should be applied to get the desired pH change.

**Step 5.** Adjust the amount of ash to apply for moisture percentage. Divide the amount of applied ash by the percentage of dry matter in the ash. In our example, if we assume that the percent moisture is 20%, then the percent dry matter is 80% (100% minus 20%). Divide 10.0 tons/acre of dry ash by 0.8 to determine that 12.5 tons/acre of ash with 20% moisture should be applied to raise the pH from 4.5 to 6.5.

Liming capacity  
Lime equivalent = 11%  
Percentage lime = 0.11

CEC of 6 = sand or loamy sand

Change in pH desired in 6" plow depth layer = 4.5 to 6.5  
Tons of lime required for different soil texture = Sand =  
1.1 tons/acre

$1.1 \text{ tons/acre} \div 0.11 = 10.0 \text{ tons/acre of dry ash}$

$10 \text{ tons/acre} \div 0.8 \text{ (percentage dry matter)} =$   
 $12.5 \text{ tons/acre of moist ash}$

# Nitrogen...

- This is a tough question. The general rule is that pasture will take up about a pound of nitrogen per acre per day (about 150 lbs/year)
- Pasture gets nitrogen from a variety of sources including legumes
- Tissue tests will nearly always show a deficiency in nitrogen and pasture will nearly always respond to an application of it—the question is generally economic



# Sulfur...

- This is an easier question. Book values show sulfur needs to be added tissue tests indicate there is less than .10%

# Phosphorus...

- This is an easier question. Book values show phosphorus needs to be added soil tests indicate there is less than 20 ppm present

# Remember this slide?

**Table 6.** Interpreting the results of soil tests and rates of fertilizer required to amend deficient soils

| Nutrient                                     | If soil test is . . . *                                | Suggested fertilizer rate  |
|--|--|--|
| Phosphorus<br>(HCO <sub>3</sub> extractable) | < 5 ppm<br>5–10 ppm<br>10–20 ppm<br>> 20 ppm           | 100 lb P <sub>2</sub> O <sub>5</sub> /acre<br>50 lb P <sub>2</sub> O <sub>5</sub> /acre<br>25 lb P <sub>2</sub> O <sub>5</sub> /acre<br>none |
| Potassium<br>(ammonium acetate extractable)  | < 40 ppm<br>40–60 ppm<br>> 60 ppm                      | 200 lb K <sub>2</sub> O/acre<br>100 lb K <sub>2</sub> O/acre<br>0–50 lb K <sub>2</sub> O/acre  |
| Zinc (DTPA extractable)                      | < 0.5 ppm (soil pH < 7.0)<br>< 0.5 ppm (soil pH > 7.0) | 5 lb Zn as ZnSO <sub>4</sub> /acre<br>10 lb Zn as ZnSO <sub>4</sub> /acre  |

\* Source: Soil and Plant Tissue Testing in California (UC ANR Bulletin 1879).

Based on our soil test, assume we need to add 50 lbs of P205 equivalent/acre

- How much 11:52/acre do I need to apply to achieve this?
  - Answer-about 100 lbs
- Assuming 11:52 costs \$660/ton, what is the per acre cost?
  - Answer-about \$33/acre
- At the 100 lbs rate, how many units of N were applied?
  - Answer-About 11 lbs

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  - Answer-about \$33/acre
- At the 100 lbs rate, how many units of N were applied?
  - Answer-About 11 lbs

Based on our soil test, assume we need to add 50 lbs of P<sub>205</sub> equivalent/acre

- How much 16:20/acre do I need to apply to achieve this?
  - Answer-about 250 lbs
- Assuming 16:20 costs \$416/ton, what is the per acre cost?
  - Answer-about \$52/acre
- How many units of N are being applied at this rate?
  - Answer-about 40 lbs

# Potassium...

- This is an easier question. Book values show potassium needs to be added if soil tests indicate there is less than 40 ppm present
  - Sources of Potassium include:
    - Potassium chloride-KCl-0-62-0
    - Potassium nitrate-KNO<sub>3</sub>-13-45

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| Potassium<br>(ammonium acetate extractable)  | < 40 ppm<br>40–60 ppm<br>> 60 ppm                      | 200 lb K <sub>2</sub> O/acre<br>100 lb K <sub>2</sub> O/acre<br>0–50 lb K <sub>2</sub> O/acre  |
| Zinc (DTPA extractable)                      | < 0.5 ppm (soil pH < 7.0)<br>< 0.5 ppm (soil pH > 7.0) | 5 lb Zn as ZnSO <sub>4</sub> /acre<br>10 lb Zn as ZnSO <sub>4</sub> /acre  |

\* Source: Soil and Plant Tissue Testing in California (UC ANR Bulletin 1879).



Based on our soil test, assume we need to add 50 lbs of K<sub>2</sub>O equivalent/acre

- How much KCl (60%) do I need to apply to achieve this?
  - Answer-about 100 lbs
- Assuming KCl costs \$822/ton, what is the per acre cost?
  - Answer-about \$41/acre
- At the 100 lbs rate, how many units of N were applied?
  - Answer-About 0 lbs

## Six questions to ask yourself before you add fertilizer:

1. Which elements do I need? (N, P, K, S, Ca)
2. How much do I apply?
3. What type of material do I use?
4. Which application method is best?
5. When is the best time to apply it?
6. Will I get a return on my investment?

## What about Manure: Relative values of manure...

| Table 3b. Approximate Nutrient Composition of Selected Types of Manure at Time of Application:<br>Type of Manure | Moisture Content % | Total N        | NH <sub>4</sub> -N | P <sub>2</sub> O <sub>5</sub> | K <sub>2</sub> O |
|--|--------------------|----------------|--------------------|-------------------------------|------------------|
|  |                    | Pounds per Ton |                    |                               |                  |
| Swine  | 82                 | 10             | 6                  | 9                             | 8                |
| Beef   | 32                 | 23             | 7                  | 24                            | 41               |
| Dairy Cattle   | 46                 | 13/0.65% DM    | 5/0.25% DM         | 16/0.80% DM                   | 34/1.7% DM       |
| Sheep  | 31                 | 29             | 5                  | 26                            | 38               |
| Chicken w/o litter   | 55                 | 33             | 26                 | 48                            | 34               |
| Turkey w/o litter  | 78                 | 27             | 17                 | 20                            | 17               |
| Horse w/o bedding  | 22                 | 19             | 4                  | 14                            | 36               |

1 These values are derived from the USDA, SCS, Agricultural Waste Management Field Handbook (1992), and modified with data collected from Colorado feeding operations when possible. Nutrient composition of manure will vary with age, breed, feed rations and manure handling practices.



## What about some other organic sources?

| Approximate Nutrient<br>Composition of Selected<br>Types of fertilizers | Moisture<br>Content % | Total N        | NH <sub>4</sub> -N | P <sub>2</sub> O <sub>5</sub> | K <sub>2</sub> O |
|---|-----------------------|----------------|--------------------|-------------------------------|------------------|
|   |                       | Pounds per Ton |                    |                               |                  |
| Cogeneration ash  | 4%                    | .29%           |                    | 1.79%                         | 1.27%            |
| Organic Dry Fertilizer  | 0                     | 9%             |                    | 3%                            | 0                |
| Guano plus  | 0                     | 11             |                    | 5                             | 2                |
| Seabird Guano   | 0                     | 12             |                    | 12                            | 2.5              |
| Rock Phosphate  | 0                     | 0              |                    | 3% Available                  |                  |

1 These values are derived from a variety of sources

How many pounds of each product on a dry matter basis would need to be applied to equate to 100 lbs/acre of P2O5?

| Approximate Nutrient Composition of Selected fertilizers | Moisture Content % | Total N | NH4-N | P2O5         | K2O   | How much product? |
|--|--------------------|---------|-------|--------------|-------|-------------------|
|  |                    | Percent |       |              |       |                   |
| Mono Ammonium Phosphate                                  | 0                  | 11      | N/A   | 52           | 0     | 200 lbs           |
| Dairy Manure   | 0                  | .65%    | N/A   | .80%         | 1.7%  | 12,500 lbs        |
| Cogeneration ash   | 4%                 | .29%    | N/A   | 1.79%        | 1.27% | 5,882 lbs         |
| Organic Dry Fertilizer                                   | 0                  | 9%      | N/A   | 3%           | 0     | 3,333 lbs         |
| Rock Phosphate   | 0                  | 0       | N/A   | 3% Available |       | 3,333             |

1 These values are derived from a variety of sources

How many pounds of each product on a dry matter basis would need to be applied to equate to 100 lbs/acre of P2O5?\*

| Approximate Nutrient Composition of Selected fertilizers | Moisture Content % | P2O5          | How much product? | Cost/Ton | Cost/Acre |
|--|--------------------|---------------|-------------------|----------|-----------|
|  |                    |               |                   |          |           |
| Mono Ammonium Phosphate                                  | 0                  | 52            | 200 lbs           | \$660    | \$66      |
| Dairy Manure   | 0                  | .80%          | 12,500 lbs        | Free?    | 0?        |
| Cogeneration ash   | 4%                 | 1.79%         | 5,882 lbs         | Free?    | 0?        |
| Guano plus   | 0                  | 5%            | 2000 lbs          | \$1600   | \$1600    |
| Sea Bird Guano   | 0                  | 12% Available | 833               | \$997    | \$415     |

\*Does not include application cost

# Fertilizing with manures

- Good source of nutrients and organic matter
- Protects soil from raindrop impact and erosion
- Local supply often available at no cost





# Different sources of manure have different amounts of nutrients

- Poultry manure is high in nitrogen
- Beef and horse manure tends to be lower in nitrogen
- Sheep manure is high in potassium (K)



UNCE, Reno, NV

# Cautions for fertilizing with manures

- Watch out for weeds and pathogens
- Know the manure nutrient content
- Consider the salt content



# Cautions for fertilizing with manures

- Incorporate or apply evenly to avoid smothering plants
- Don't apply on frozen slopes
- Avoid leaching nutrients into waterways



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## Six questions to ask yourself before you add fertilizer:

1. Which elements do I need? (N, P, K, S, Ca)
2. How much do I apply?
3. What type of material do I use?
4. Which application method is best?
5. When is the best time to apply it?
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