

Guidelines for interpreting laboratory data on the suitability of irrigation water for
SALT SENSITIVE DECIDUOUS TREE CROPS
 (Almond, Apple, Apricot, Cherry, Peach, Pear, Walnut)
 (acceptable range for pH: between 6.5 – 8.4)

The Problem and related constituents	No Problem	Increasing Problem	Severe problem ⁵
Salinity: stunts tree growth & reduces yield ECw ¹ (dS/m or mmhos/cm) ECw ¹ (uS/cm) TDS ² (mg/L or ppm)	<1.1 1100 <700	1.1– 3.2 1100 – 3200 700 - 2000	>3.2 > 3200 > 2000
Permeability: affects the rate of water movement into and through soil. When: SAR ³ = 0-3 and ECw = SAR = 3-6 and ECw = SAR = 6-12 and ECw = SAR = 12-20 and ECw = SAR = 20-40 and ECw =	> 0.7 > 1.2 > 1.9 > 2.9 > 5.0	0.7 - 0.2 1.2 – 0.3 1.9 – 0.5 2.9 – 1.3 5.0 – 2.9	< 0.2 < 0.3 < 0.5 < 1.3 < 2.9
Toxicity⁴: specific ions that can injure and affect tree growth during surface irrigation Sodium (meq/l) Sodium (ppm or mg/l) Chloride (meq/l) Chloride (ppm or mg/l) Boron (ppm or mg/l)	< 3.0 < 70 < 4 < 140 < 0.5	3.0 – 9.0 70 - 200 4 - 10 140 - 350 0.5 - 3	>9 >200 >10 > 350 > 3
<p>Notes:</p> <p>¹ ECw = Electrical Conductivity of water. This is a general measure of the overall salinity of water. It is typically reported in units of milli mhos per centimeter (mmhos/cm), deciSiemens/meter (dS/m) or microSiemens/centimeter (uS/cm) ; dS/m x 1000 = uS/cm and dS/M = mmhos/cm.</p> <p>² TDS = Total Dissolved Solids is another measure of overall salinity and is measured in (mg/L) or (ppm). <i>Conversion*</i>: Divide TDS (mg/L or ppm) by 640 to get EC (dS/m) (*this is only accurate if the EC is < 5 and the sulfates are not excessive)</p> <p>³ SAR = Sodium Adsorption Ratio. Calculated by the testing lab from the relative amounts of sodium, calcium, magnesium in water: $SAR = Na^+ / \sqrt{(Ca^{2+} + Mg^{2+}) / 2}$</p> <p>⁴ Individual ions may be reported either as milliequivalents per liter (meq/l) or parts per million (ppm); Parts per million is the same as milligrams per liter (mg/l)</p> <p>⁵ Special management practices and favorable soil conditions are required to successfully produce trees with water of this quality. Not suitable for typical agricultural use.</p> <p>Include the nitrogen applied with your irrigation water when planning your fertilizer program so you don't apply too much. Multiply nitrate-nitrogen by 2.7 to determine the pounds of nitrogen per acre-foot of water.</p>			
<p><i>Excerpted from: Agricultural Salinity & Drainage, University of California Agriculture & Natural Resources Publication 3375, 2006 – available from: http://anrcatalog.ucdavis.edu</i></p>			

Other UC Resources:

- **Irrigation Water Salinity and Crop Production**, University of California Agriculture & Natural Resources Publication 8066; download for free from <http://anrcatalog.ucdavis.edu>
- **UC Drought Management Website:** <http://ucmanagedrought.ucdavis.edu/>
- **CA Institute for Water Resources Website:** <http://ciwr.ucanr.edu>