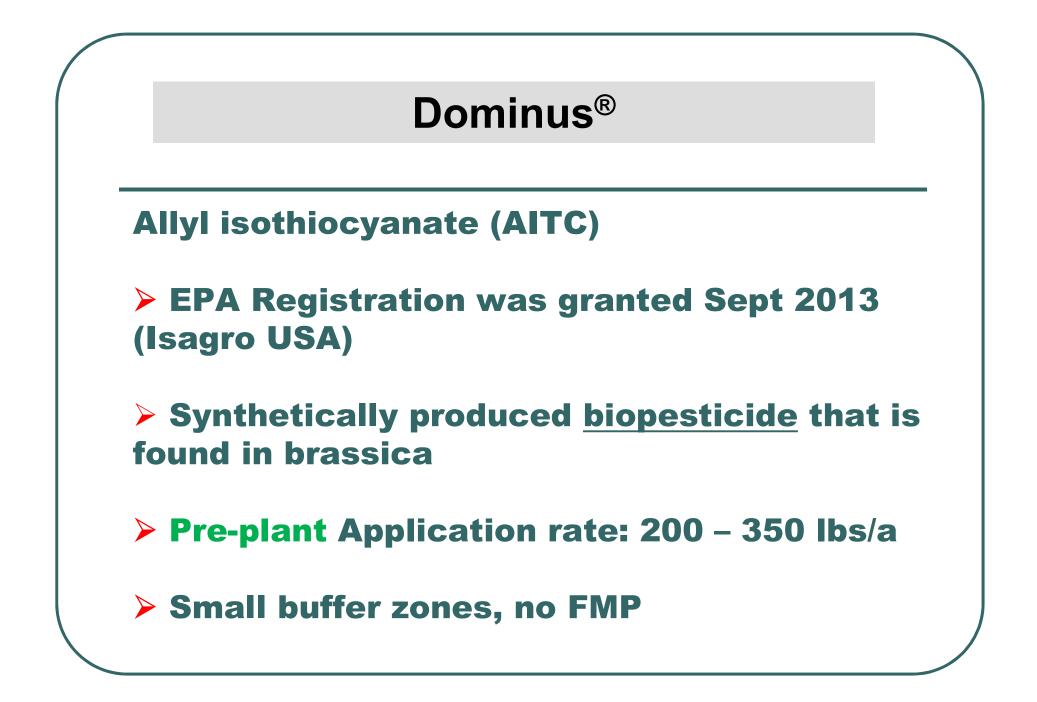
New Fumigants for Strawberry Production in California

Dominus[®], Paladin[®], Trifecta[®], and EDN[®]

Husein Ajwa

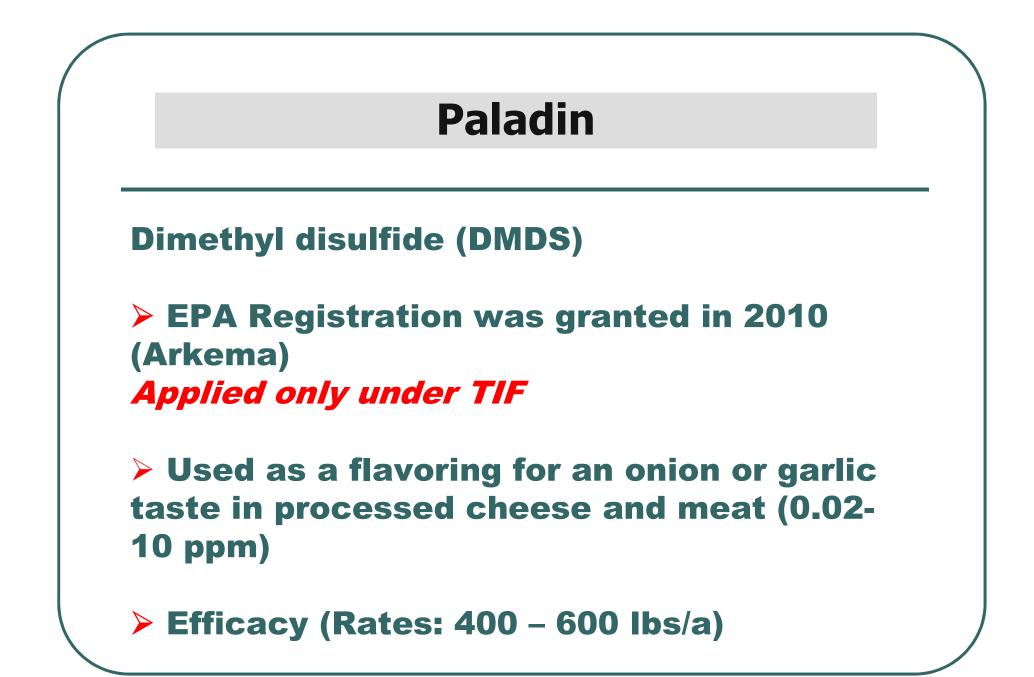
UCCE Emeritus





Selected properties of Dominus

CAS #	57-06-7	
Boiling point	151°C	
Vapor pressure	3.5 – 4 mm Hg	
Vapor density (water = 1.0)	3.4	
Density	1.0126 g/cm ³ @ 20°C	
Solubility in water	Slight, 2 g/L water @ 25°C	
Solubility in alcohol	Very soluble (1:8) in 80% ethanol	
Henry's Law Constant	0.0002752 atm-m ³ /mole	
Molecular formula	C ₄ H ₅ NS	
Molecular weight	99.1542 g/mol	
Soil half-life, DT50)	Aerobic DT50 of less than 3 days	



Laboratory Dose-Response Studies









Laboratory Dose-Response Studies

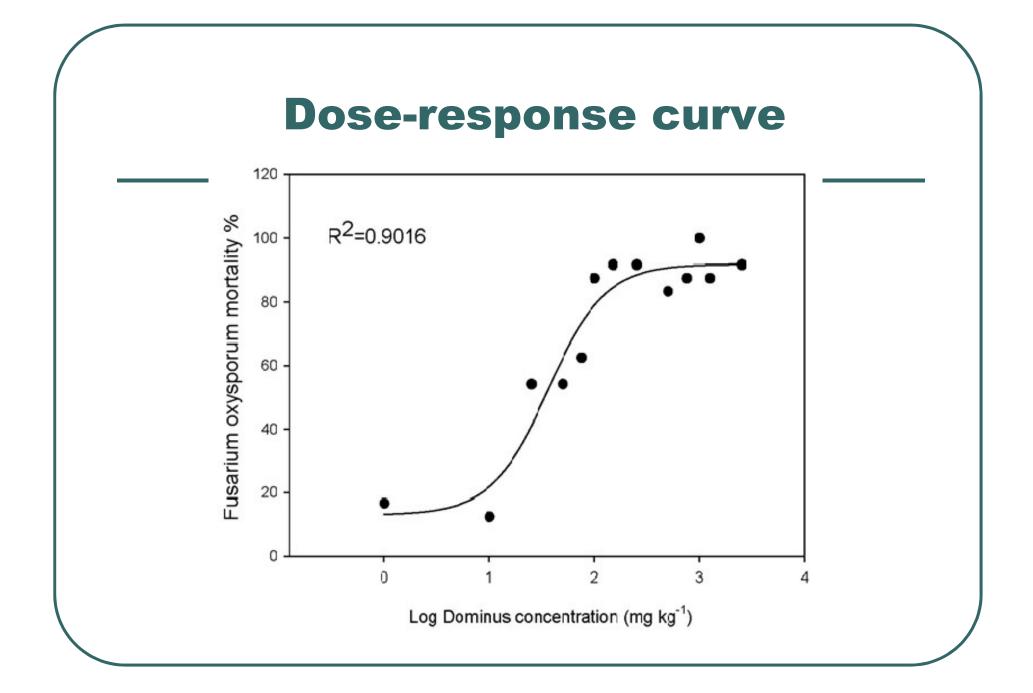


Dominus was injected as liquid with syringes and the jars were immediately sealed Dominus doses: 0, 10, 25, 50, 100, 150, 250, 500, 750, 1000, 1250, 2500 ppm

REPIPET®

The fumigated jars were placed in incubators at 20°C for 24 hours





Dose (lbs/ac) required to control 90% of pathogen population (LD₉₀)

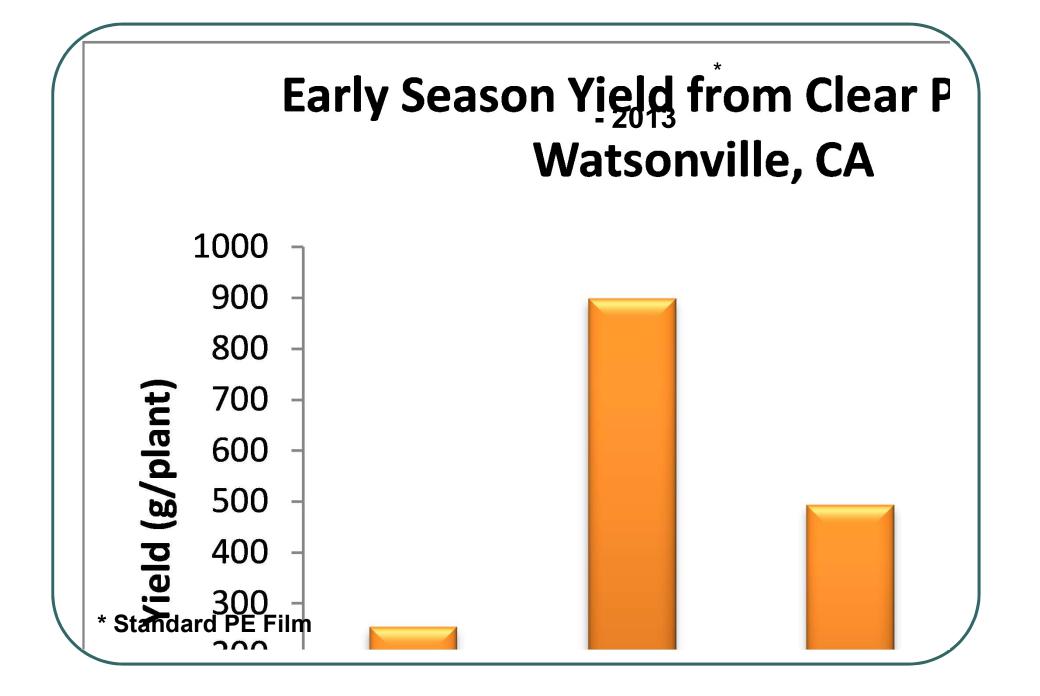
Fumigant	Fusarium oxysporum	Verticillium dahliae
Paladin (79:21)	315	406
Dominus (96%)	232	159
PicClor 60 (60:40)	146	230

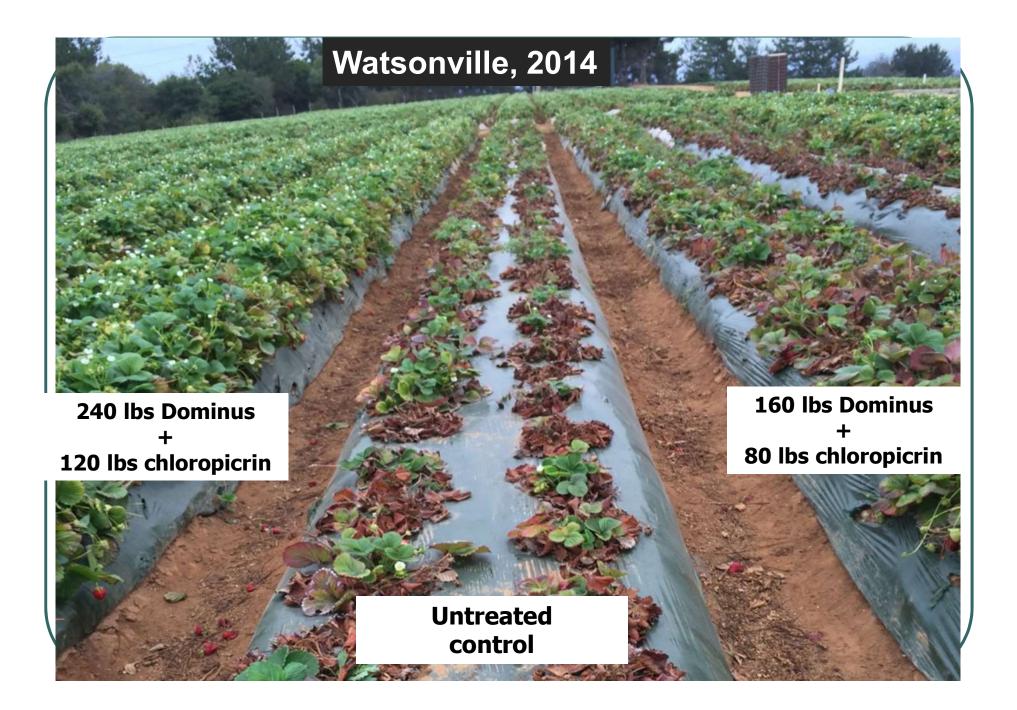
1 mg/kg = 2 lbs/ac for a soil depth of 6.7 inches

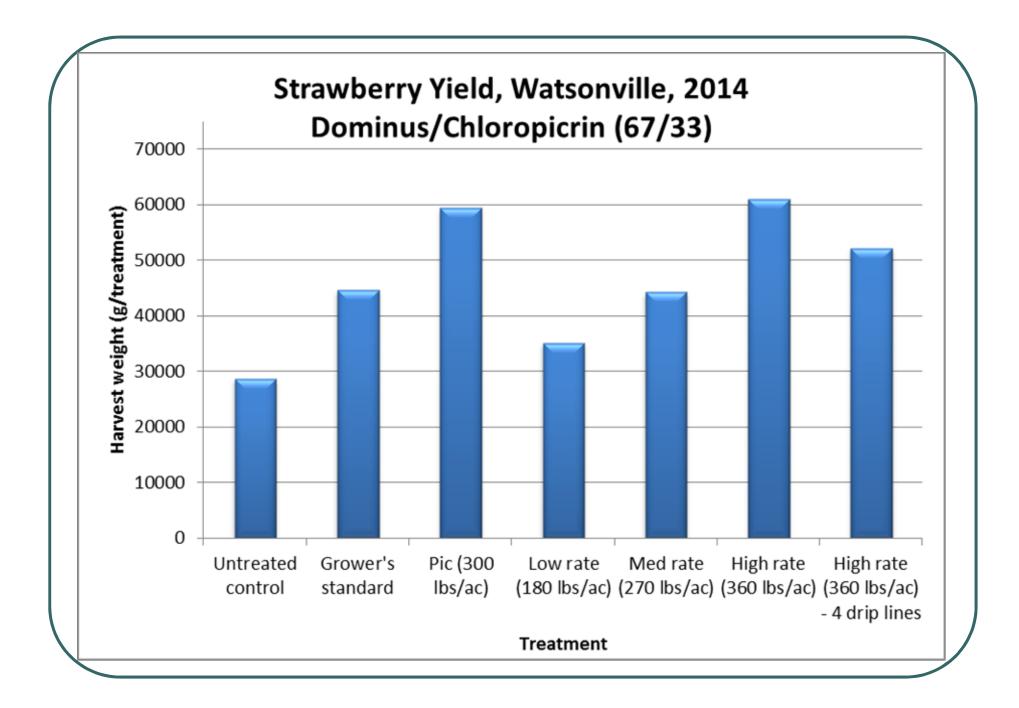
Field Research

Watsonville: Non-fumigated field for past 2 years. *Fusarium* pressure is severe.









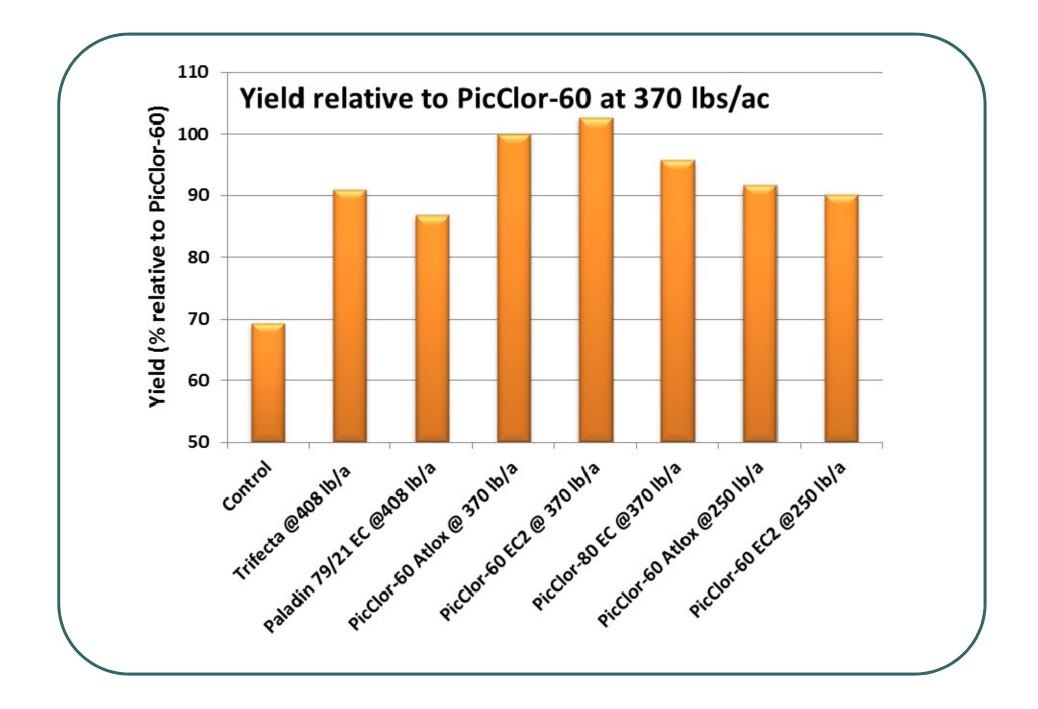
DRIP FUMIGATION Dominus Treatments, 2014-2015

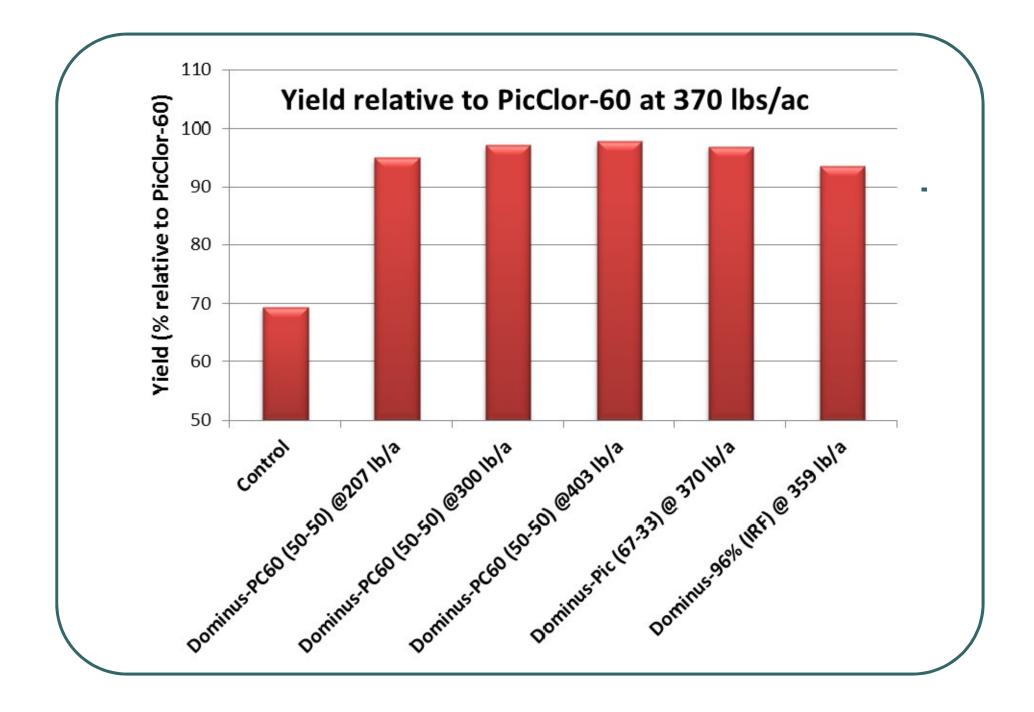
Treatment	Rate (Ibs/ac)	
Dominus/PicClor-60 (50/50)	207	
Dominus/PicClor-60 (50/50)	300	
Dominus/PicClor-60 (50/50)	403	
Dominus/Chloropicrin (67/33)	370	
Dominus (96%)	359	

Paladin and PicClor-80, 2014-2015

Treatment	Rate
PicClor-80 (80% chloropicrin)	370 gal/ac
Paladin 79:21	408 lbs/ac
Trifecta (DMDS/Pic/1,3-D)	408 lbs/ac

Rates are per treated acre. Broadcast equivalent = treated acre rate x 0.63





Summary Dominus (IRF-135)

- No phytotoxicity or plant injury was observed when planting 10 days after fumigation.
- Dominus alone or with Chloropicrin (50/50) produced strawberry yields >94% compared to the standard PicClor-60 EC at 370 lbs/ac.

2016 research

- to evaluate optimum amount of irrigation water for drip fumigation of a new emulsifiable concentrate formulation of Dominus for strawberry production in California.
- to evaluate the effect of an azeotrope on the mobility of Dominus in soil.
 - The focus of this presentation is to evaluate the overall yields during the growing season of 2015-2016.

Dominus Treatments, 2015-2016

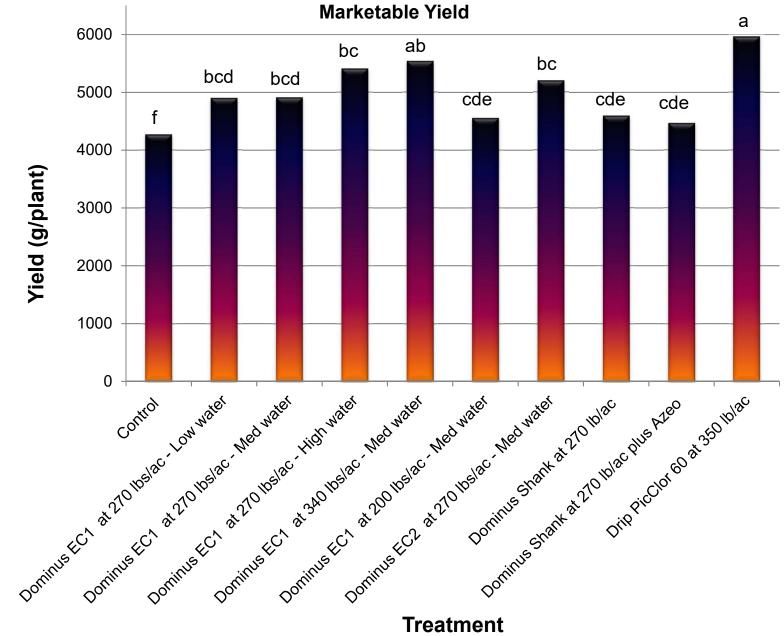
DOMINUS EC1- Low water (0.6")	270 lb/ac
DOMINUS EC1- Med water (1.0")	270 lb/ac
DOMINUS EC1- High water (1.4")	270 lb/ac
DOMINUS EC1- Med water (1.0")	340 lb/ac
DOMINUS EC1- Med water (1.0)	200 lb/ac
DOMINUS EC2- Med water (1.0")	270 lb/ac
DOMINUS Shank	270 lb/ac
DOMINUS- Shank plus Azeotrope	270 lb/ac
Drip PicClor-60 EC	350 lb/ac

Shank injection: 2 shanks spaced 10 inches apart

AITC and a fluorocarbone azeotrope mixture was applied at a 50/50 ratio to soil (270 lbs/ac of AITC) using 2 shanks/bed at 8" deep.







Summary Dominus (IRF-135)

- Strawberry yields increased with increasing the amount of water used to apply Dominus.
- Adding an Azeotrope to Dominus did not improve Dominus diffusion in the soil.
- A new emulsifier and other azeotropic combinations are being evaluated.

FATE OF ETHANEDINITRILE (C₂N₂) FUMIGANT IN SOIL

⇒ EDN[™] FUMIGAS manual for fumigation



EDNTM FUMIGAS fumigant. Manual for fumigation.



Selected Properties of Ethanedinitrile N=C-C=N

Product Brand Name:	EDN (ETHANEDINITRILE, C ₂ N ₂)
USEPA Reg. No.:	62719-321
% Active Ingredient:	99.58%
Chemical Family:	DiCyanogen
Color, Odor.	Colorless gas, almond-like odor.
Molecular Formula:	C_2N_2
Molecular Weight:	52
CAS No.:	460-19-5
Density:	Gas: 2.189 mg/cm ³ @ 20°C
	Liquid: 989 mg/cm ³ @ -40°C
Boiling Point:	-20°C
Vapor Pressure:	5.16 bar @ 21.1 ⁰ C
Solubility in water	450 cm ³ /100 cm ³ water
@101.325kPa@20°C	

Properties of EDN

- > It diffuses through soils quickly.
- Threshold Limit Value (TLV Human) = 10 ppm or 21 mg/m³.
- \succ LC₅₀ (inhalation) 350 ppm/1 hour (rat).
- LDL_o (subcutaneous) 13 mg/kg (rabbit).
- It is effective in controlling soil-borne fungal pathogens, nematodes, and many weeds.

PRELIMINARY INVESTIGATION OF ETHANEDINITRILE FOR CONTROL OF WEEDS AND NEMATODES IMPORTANT TO FLORIDA PRODUCTION SYSTEMS

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Background

Previous experiments conducted in Australia with ethanedinitrile have demonstrated control of weeds and diseases of importance to the production of a variety of crops (Ren et al., 2002). Pests included seed and soilborne fungi (Smith et al., 2003) as well as plant-parasitic nematodes. Weed control in these trials was most effective when plots were tarped and was dependent upon species (Mattner et al., 2003; Ren et al., 2003).

Methods

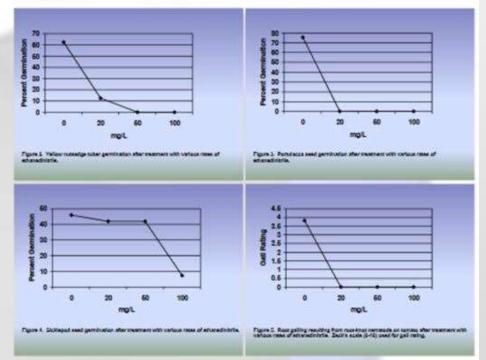
A preliminary in vitro experiment was conducted with seeds of several weed species of importance in vegetable and ornamental production systems in Florida, and with root-knot nematode (Meloidogyne incognita) infested soil.

The prepared weed and nematode inoculum were placed in open desiccators of measured volume, allowed to equilibrate to the test relative humidity, sealed, and injected with a test amount of EDN through a gas septum port, having first withdrawn an equivalent volume of air (Figure 1).





Results

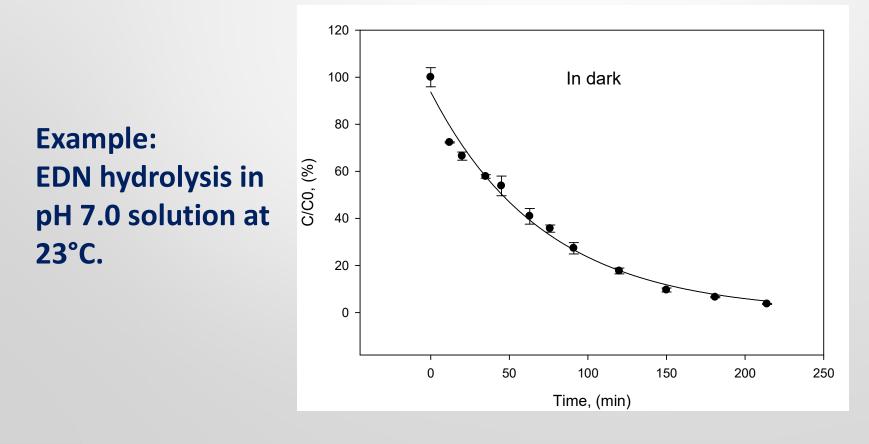


Research Objectives:

- 1. To evaluate the persistence (degradation) of EDN in different soil types.
- 2. To measure the production and disappearance of EDN's main metabolites in different soil types.

EDN degrades very rapidly in air, soil, and water. The half-life:

- in air: 100 days (light) to 150 days (dark).
- in soil and water: minutes to days, depending on the pH and temperature.



EDN degradation in water (hydrolysis) as a function of pH and temperature

pH/ Temp. (°C)	рН 4.0	pH 7.0	рН 9.0
10	80 *days	257 min	6.0 min
23	28 [#] days	49 min	4.5 min
40	1.5 days	11 min	1.7 min

Summary

- Application rates might need to be adjusted based on soil pH and temperature.
- EDN metabolites are expected to degrade to nontoxic compound within days after fumigation.
- Possible degradation products:
 - NCCN + $2H_2O \rightarrow HCN + HOCN$
 - HOCN + $H_2O \rightarrow NH_3 + CO_2$
 - HCN + $H_2O \rightarrow HC(O)NH_2 + H_2O \rightarrow HCO_2^- + NH_4^+$
- HCN converts into thiocyanate (SCN) and eventually into SO₄ + NH₃ + CO₂ or forms precipitates with metals (eg., Fe)

