

## Potassium: Use and Application and Availability in Soils

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## Plant Physiology of Potassium

- Activator of enzymes
- Involved in stomatal function
- Helps to regulate ion balances in cells
- Involved in sugar synthesis
- Increases oil content in pistachio fruit
- Contributes to cold hardiness

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## Potassium Deficiency

- Leaves become pale during summer
- Leaflets fold upward and curl in
- Yellow tips that progress inward; tissue eventually browns
- Slow growth
- Small leaves
- Low yield



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## Physiology of Potassium

- Most potassium needed by bearing trees is to supply the fruit
- Most of K uptake in bearing trees is for kernel fill
- 29 lbs  $K_2O$  removed per 1000 lbs of nuts removed
- 27 lbs  $K_2O$  needed to supply tree growth

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## Leaf analysis levels

Critical Value	Adequate
1.6%	1.8 – 2.2%

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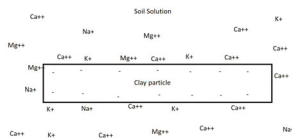
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## Potassium in Soils

- Potassium is immobile in soils because it is loosely bound onto negatively charged clay and organic matter
- Clay and organic matter forms the "CEC"
- The size of your CEC is dependent on how much clay and organic matter you have




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## Cation Exchange Capacity

- The CEC acts as a reservoir for positively charged ions
  - K, Ca, Mg? Good!
  - Na: bad!
- When thinking about cations, you also must be thinking about the CEC size
- Lab tests will give you ppm K, %K on the CEC, or both
  - %K is usually more useful, since it gives you an idea of how much of the CEC is taken up by K
- *Rough* rule of thumb: soils are “deficient” in K with either less than 200 ppm OR 2% of the CEC
- It’s best to just apply what has been lost via crop removal

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## Potassium Fixing Soils

- Ammonium acetate test not appropriate for soils that fix K
- Fixed K becomes available over time
  - Rate can change based on soil
- Do not add K fertilizers far in advance of tree needs in K fixing soils

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## Cation antagonism

- Carrier proteins for nutrient uptake is not always specific to an ion
  - Worse for cations of similar charge and/or size
  - Example:  $K^+$  and  $Na^+$ ,  $K^+$  and  $NH_4^+$
  - Also  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$
  - $Cl^-$  and  $NO_3^-$
- But you can have ion antagonism in extreme cases
  - E.g. Ca deficiency in high Mg soils such as serpentine
  - Depression of one cation with application of fertilizer containing another
- Not much evidence that there is an ideal cation ‘ratio’ that is perfect across all sites
  - Maintaining this may be expensive

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## Soil Tests

- Ammonium Acetate extraction:
  - Supposed to measure available K
  - Not great at estimating available K in fixing soils
- Water soluble (saturated paste extraction):
  - Only removes water soluble K
  - No sufficiency standards developed
- AA tests will tell you ppm in soil
- Labs will calculate %K for you
- CEC in most cases is estimated
  - Can be VERY inaccurate if recently applied gypsum, lime and sampler took no steps to exclude it

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## Potassium Fertilizers

- K<sub>2</sub>O: Potassium Oxide, known as potash. Not actually in fertilizers! 83%K

Source	Formula	% K <sub>2</sub> O	Solubility@ 20 C (grams/liter)	% K <sub>2</sub> O in a saturated solution	Cost per lb K <sub>2</sub> O
Potassium Chloride	KCl	63	255	16.1	\$0.41
Potassium Sulfate	K <sub>2</sub> SO <sub>4</sub>	54	111	5.4	\$0.67
Potassium Nitrate	KNO <sub>3</sub>	47	209	11.2	\$1.33
Potassium Thiosulfate	K <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	25	complete	25	\$1.38

Solubility data may differ by manufacturer, or if the amount of impurities is high  
 Costs are quotes from 2019, and actual prices may differ

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## Previous Research

- Patrick Brown, Brent Holtz, and Quipeng Zeng looked at K fertilization in a mature pistachio orchard in a San Joaquin series soil
- They added 0, 100, 200, 300 lbs of K/acre
- Looked at three different sources: K<sub>2</sub>SO<sub>4</sub>, KCl, KNO<sub>3</sub>
- Trial lasted 3 years, with K split into 4 applications per year

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## Zeng, Brown, and Holtz.

- No difference between K source
- Yield maximized at 200 lbs/year

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## Solubility

- Tells you the maximum amount of a salt that will dissolve in pure water at a given temperature
- Solubility of potassium fertilizer does not change how potassium ultimately behaves in the soil
  - But there are tricks that can reduce the importance of fixation
- **Easily soluble fertilizer applied early in the season (or last year) in potassium fixing soils can still be fixed**
- The key to potassium fertilizer application is applying a product that works for you in a way that is *appropriate for your soil*

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## Solubility of Fertilizers and Application

- Solubility can become an issue in fertigation, particularly if you like to apply all of your K over a small number of applications
- Higher soluble fertilizers are more efficient lb/lb, (You can deliver the same amount of dissolved  $K_2O$  in less water than lower soluble fertilizers) but there are no differences in 'available' potassium

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## Fertigation vs Banding?

Treatment	1996	1997	1998	1999
No K	2192	2500	2868	2904 d
0.38 kg SOP	2382	2719	2916	3313 bc
0.75 kg SOP	2305	2797	2792	3335 abc
0.38 kg MKP	2251	2862	3067	3727 a
0.38 kg KTS	2345	2867	2824	3015 cd
0.75 kg SOP banded	2275	2978	2585	3534 ab

- Banded SOP achieved similar yields in year three to fertigated SOP (same rate)
- MKP was (not significantly) best – unknown if this was due to P or something else

Adapted from: Edstrom, J.P. and Meyer, R.D., 2006, August. Potassium fertilizer application in drip and micro-jet irrigated almonds. In *V International Symposium on Irrigation of Horticultural Crops 792* (pp. 257-263).

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## Fertigation vs soil application?

Soil depth	ppm potassium (via ammonium acetate extraction) of soils after 2.5 lbs K had been 'broadcast' under the emitter or fertigated								
	0 feet from emitter			1 foot from emitter			2 feet from emitter		
	Broadcast	Fertigated	Control	Broadcast	Fertigated	Control	Broadcast	Fertigated	control
0-6"	2131	520	270	164	512	250	164	305	289
6-12"	2714	414	141	137	395	137	141	180	98
12-18"	3284	332	113	109	258	102	121	117	70
18-24"	3288	230	74	78	407	70	78	94	94
24-30"	1634	100	70	74	228	66	74	63	106
30-36"	176	66	90	86	90	82	98	86	86

Adapted from: Uriu, K., R.M. Carlson, D.W. Henderson, H. Schulbach, and T.M. Aldrich. 1980. Potassium fertilization of prune trees under drip irrigation. *J. Amer. Soc. Hort. Sci.* 105:508-510.

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## Fertilizer Application Strategies

- Banding SOP works in any soil type, as long as it is banded in the *wetted zone* of your irrigation system
  - Because surface-applied K mostly stays put, fertilizer that can only be dissolved by rain will not be in the root zone
- In K-fixing soils: banding saturates the fixation capacities of a soil, ensuring K is available for uptake
- Fertigation can avoid fixation, *but only if it's applied in season*
- Banding can be a great addition to fertigation, especially in non-fixing soils

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## When to fertilize

- If your trees are very deficient (< 1.6% K):
  - Apply high rates and monitor leaf levels closely
  - 100-200 lbs K<sub>2</sub>O/acre overcame deficiencies in a loam soil
- If your trees are not deficient, apply to replace what's been removed
  - It's always better to maintain levels rather than remedy deficient trees

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## Maintenance fertilization rates

Yield (kernel lbs/acre)	K <sub>2</sub> O/acre
1000	55
2000	84
3000	113
4000	142
5000	171

29 lbs K<sub>2</sub>O removed with every 1000 lbs kernels

27 lbs K<sub>2</sub>O needed for tree growth

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## When to fertilize

- Non-potassium fixing soils
  - Apply whenever you like, however you like
  - Band? Make sure it's in the wetted zone
- Potassium fixing soils
  - Fertigation: start a little before trees need it (nut fill/Stage 3), split applications between April-August
  - If soil applying, **BAND** to overcome fixation, and ensure the fertilizer is in the wetted zone
  - You can do both!

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Thank you!

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